

# USE OF A MULTISCALE/MULTIPHYSICS APPROACH FOR GEN 3B LI-ION BATTERIES MODELLING IN A SIMULATION TOOLCHAIN 31<sup>ST</sup> ISE TOPICAL MEETING - AACHEN

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#### Next generation batteries

ifpen

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This project has received funding from the European Union's Horizon 2020 research and innovation program under grant agreement No 875193.

G	Current mmercial Li-ion en 1-2b • 90-235 Wh/kg • 200-630 Wh/l	Advanced Li-ion Gen 3b	350 Wh/kg, 750 Wh/l • 2025 Solid State Gen 4 • 500 Wh/kg, 1000 Wh/l • 2025-2030		
Nationale Plattfor	rm Elektromobilität, Deutschland Jan. 2016 Cell chemistry		New technologies relying on new materials to improve batteries performances		
Generation 5	<ul><li>Li/O2 (lithium-air)</li><li>Conversion materials (LiS)</li></ul>		<ul> <li>Gen 3b</li> </ul>		
Generation 4	<ul> <li>All-solid-state batteries</li> <li>Gr ou Si/Gr anode (Gen 4a)</li> <li>Li metal (Gen 4b)</li> </ul>	> 2025	<ul> <li>Use of Silicon in the negative</li> <li>Graphite 350 mAh/g → + 15% vol</li> <li>Silicon 3000 mAh/g → +300% vol</li> </ul>		
Generation 3b	<ul> <li>Cathode: HE-NMC, High Voltage Spinnel</li> <li>Anode: Silicon/carbon (&gt;20% Si)</li> </ul>	≈ 2025	<ul> <li>Need to account for volumetric expansion</li> <li>Long term effect on aging</li> </ul>		
Generation 3a	<ul> <li>Cathode: NMC 622 to 811</li> <li>Anode: Silicon/carbon (5-10% Si)</li> </ul>		<ul> <li>Cell swelling</li> <li>Gen 4</li> <li>Use of solid electrolyte</li> <li>New ionic conduction type</li> <li>Solid/solid interfaces</li> </ul>		
Generation 2b	<ul><li>Cathode: NMC 532 to 622</li><li>Anode: 100% carbon</li></ul>				
Generation 2a	<ul><li>Cathode: NMC 111</li><li>Anode: 100% carbon</li></ul>	Current			
Generation 1	<ul><li>Cathode: LFP or NCA</li><li>Anode: 100% carbon</li></ul>		Stability issues at the Li/electrolyte interfa		





# Multi-scale/Multiphysics approach



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- All skill needed for battery design and manufacturing
  - O Modelling from atoms to system
  - Experimentation from material to system
  - Material manufacturer
  - Battery manufacturer
- All value chain for battery
  - Academics and R&I actors
  - SME
  - Manufacturing from material to cells: European Battery alliance
    - Saft
    - Solvay
    - O Umicore
    - Siemens AG
  - OEM

### **Expected achievement: Seamless interfaces from nano to macro scales and from ns to years**



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#### Specific needs from all stakeholders



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#### Gen 3b toolchain development



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Each scale brings information necessary for understanding and forecasting the complete cell behavior







umicore

materials for a better life

SOLVAY

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- Gen 3b cell
  - O NMC811
  - SiOx/C negative electrode
  - 1M LiPF6 in EC:DMC (3:7) + 10% FEC
- Commercial negative material
  - Complex formulation needing specific characterization
    - Graphite particles: 77%
    - Si based particles
      - Si inclusion in SiO2 matrix + Graphite layer









# Intrisic material properties through ab initio modelling



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- Simulation focused on intrinsic properties for the material
  - Diffusion properties investigated through Nudged Elastic Band methodology
    - Evaluation of hoping *a* distance between adjacence sites

$$D(T) = a^2 v^* \exp\left(-\frac{\Delta G^{++}}{KT}\right)$$

- $\Delta G^{++}$  Gibbs free energy of activation for ion hopping calculated by NEB
- $\nu^*$  effective attempt frequency calculated by phonon calculations at the beginning and transition states

Case	∆G <sup>++</sup> (eV)	ν* (Hz)	D(cm²/s) @T=300K
ODH	0.78	3.5E+13	2.6E-15
TSH	0.56	3.2E+12	1.6E-12

- Same methodology applied in solid state electrolytes
- OFT modelling of negative electrode active materials
  - Ab initio molecular dynamics
    - Creation of amorphous silicon and amorphous lithium silicide models
    - Interpolation of volumetric expansion vs. Li content curve
    - Analysis of bond-breaking patterns at low and high Li contents
  - Elastic properties:
    - Ab-initio calculation of elastic tensor and mechanical parameters
    - Consequences of lithiation and delithiation on the chemical and mechanical properties of all anode components
- Parameters can be further used in larger scale simulations





Lithiated anode materials







# Understanding of particle behaviour



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- Study of the effect of Li insertion on  $\bigcirc$ particle behaviour
  - Volume variation
  - Stress generation  $\bigcirc$
  - Mechanical properties variation  $\bigcirc$
- Uncertainties remaining  $\bigcirc$

Der

- Composition of the matrix  $\bigcirc$
- Amount of Si in the complete particle  $\bigcirc$









## Mechanical behavior of electrode composite



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- DEM model of electrode material  $\bigcirc$ 
  - Si/SiOx and C particles represented as an assembly sphere
  - Special bonding contact laws account for the presence of binder
- "One-way" coupling for electrochemical effect on swelling  $\bigcirc$ 
  - Particle volume as a function of SOC is provided by other models
  - Particle swelling as a function of time is scripted  $\bigcirc$
- Model variables
  - Initial density  $\bigcirc$
  - Pressure on the anode
  - Binder flexibility  $\bigcirc$
  - **Binder dispersion**  $\bigcirc$
- Model output  $\bigcirc$ 
  - Evolution of anode density/volume
  - Binder damage at first charge (no cycling)  $\bigcirc$
- First interesting results  $\bigcirc$ 
  - No observed influence of binder properties on swelling behaviour
  - Binder flexibility is the key property that pilot binder damage  $\bigcirc$
  - High pressure on electrode collapse porosity and damage binder  $\bigcirc$











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Start pressure Start swelling

0.68

0.66

0.6

0,62 Solid

0,6

0,58

0.56 0.5

-0.2

Impact of initial density (pressure : 1 MPa)

0.4

Relative time

-0,55 -0,59 -0,64

0.2





- Creation of a representative electrode volume using DEM or stochastic approaches
- Coupled simulation of mechanical behavior and electrochemical behavior
  - Charge behavior of particles depends
    - Their geometry
    - Their nature (Si based or graphite)
  - Evaluation of electrode variation over cycling
    - Electrode displacement
    - Endured pressure in binder/electrolyte domain
- Remaining challenges:
  - Fully coupled model based on large deformation theory



















- Use of a simplified electrochemical model
- SPM-e model with blend negative electrode
  - Compatible with long term simulation for aging
  - Accounting for nominal electrochemical behaviour
  - Accounting for aging phenomena
    - SEI layer formation on both material in the negative
    - Li plating
    - Loss of active material at the positive
  - With mechanical aspects
    - Stress induced diffusion
    - Cracks induced SEI formation
    - Loss of active material due to mechanical stresses









# Modelling of aging phenomena



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- SEI formation
  - Simplified SEI formation formulation without SEI convection
  - SEI formation current computed
    - Leading to Loss of Li inventory
    - SEI thickness increase
      - SEI formation passivation
      - Resistance increase
- Li plating
  - Reversible Li plating
  - O Lithium trapping into inactive materials → dead Li
- Loss of positive active material
  - Oxidation of positive active material leading to loss of active material
- Cracks induced SEI formation
  - Based on stresses computed in particles

$$\frac{dA_{cr}}{dt} = k_{cr} \frac{d\sigma}{dt}$$

- Increased SEI formation rate
- Cracks induced LAM

■ Based on contact losses computed in DEM → loss of active material

• 
$$\frac{d\Delta\varepsilon}{dt} = k_{LAM} \frac{dC_{loss}}{dt} \Big|_{SOC,T}$$





Charge balance leading to loss of Li inventory

$$\begin{split} I_{neg} &= I_{gr} + I_{SiOx} + I_{SEI} + I_{pl} \\ I_{pos} &= I_{NMC} + I_{ox} \\ \text{Impact of loss of active material} \\ \varepsilon^{s}_{neg} &= \varepsilon^{s}_{neg}_{0} (1 - \Delta \varepsilon_{LAM}) \\ \varepsilon^{s}_{pos} &= \varepsilon^{s}_{pos}_{0} (1 - \Sigma \Delta \varepsilon_{LAM}) \end{split}$$



### Model outputs



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- Classical cell behaviour
  - Voltage
  - Heat release
  - Swelling
    - From DEM calculation
      - Cell length change given by material volume change
    - 2 steps during discharge
      - Gr deintercalation low volumetric change
      - Si deintercalation with high volumetric change
- Non accessible variables

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- Material concentrations
- Ourrent repartition between materials
- Potentials







# Evaluation of cell formation



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- Aim simulate a realistic formation cycle
  - Pristine electrode are set-up in the model
    - Fully lithiated NMC in the positive
    - Empty negative electrode
  - Formation cycle used for real MODALIS<sup>2</sup> cells
  - Oesign taken from Saft
  - Model operates as a test bench and automatically switch between rest, CC and CV
- 41 h simulated in 10 minutes
- Initial cell capacity computed
  - 5,58 Ah VS 5,63 as mean capacity for manufactured cells







### • CC/CV cycles simulated

- From initial formation starting point
- Fast simulation (7 parallel runs)
  - 400 000 s simulated (36 full DoD cycles)
    - 3 000 s computational time
  - 800 000 s simulated (57 full DoD cycles)
    - 4300 s computational time
- Evaluation of the effect of each aging phenomenon
- Cycle:
  - CC at 6A or 3 A then CV at 4,2 V
  - O CC at -6 A until 2,7 V
  - O Cell capacity evaluated at 6A during discharge
- Main aging phenomena (qualitatively)
  - LAM due to stresses
  - O Then SEI with visible impact of cracking







Cnergies nouvelles



SIEMFN

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Ingenuity for life Non-uniform aging in  $\bigcirc$ thermo-electro-chemically (De-)Interclalation current  $j_{\rm neg}(c_{\rm neg}, \varepsilon_{\rm active}, A_{\rm crack}, \delta_{\rm SEI,...})$ coupled macrohomogenous and side reaction current  $j_{\text{SEI}}(c_{\text{solv}}, \varepsilon_{\text{active}}, A_{\text{crack}}, \delta_{\text{SEI,...}})$ as sources/sinks in macrohomogeneous balance model (P2D) with realistic geometry Particle boundary conditions: Solve 1D problem and  $\frac{\partial c_{\text{neg}}(r,t)}{\partial r}$  $-\frac{J_{\text{neg}}}{nF}$ update variables: D<sub>neg</sub>. J<sub>SEI</sub> nF  $c_{\rm neg}, \varepsilon_{\rm active}, A_{\rm crack}, \delta_{\rm SEI}$ r=RTo be presented in 32<sup>nd</sup> ISE  $\bigcirc$ topical meeting





#### Conclusions and perspectives



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- Full toolchain implemented for Gen3b cells
  - Link between material electrode full cell behaviour
  - Phenomena studied
    - Mechanical behaviour
    - Diffusion
    - Aging
    - Thermal stability
- Tool chain to be validated against experimental data
  - Extensive aging campaign
  - Dedicated devices to evaluate swelling
- Future developments to focus on Gen4 cells
  - Solid electrolyes with metal Li
  - Oevelopment initiated on argyrodite behaviour
    - Li conductivity
    - Chemical stability
    - 1 D model to be adapted to account for Li electrode









M. PETIT - 31st ISE topical conference



Lab cell



**2 mAh** / 0,4 cm²

High operating pressure (>100 MPa)





# THANKS A LOT FOR YOUR ATTENTION

# HTTPS://MODALIS2-PROJECT.EU/



